

FACULTY OF ENGINEERING
Scheme of Instruction & Examination
(AICTE Model Curriculum)
and
Syllabi
of
Four Year Degree Program of
Bachelor of Engineering(B.E)
ELECTRONICS & COMMUNICATION ENGINEERING



Issued by
Dean, Faculty of Engineering
Osmania University, Hyderabad – 500 007

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Mission and Vision of Osmania University

The Vision	<p>The Vision of the University is to generate and disseminate knowledge through a harmonious blend of ancient and modern wisdom, and to serve the society by developing in students heightened intellectual, cultural, ethical, and humane sensitivities; to foster a scientific temper, and to promote professional and technological expertise. Central to this vision is a commitment to regional and national development in consonance with our culture, heritage, and environment.</p>
The Mission	<ul style="list-style-type: none">• To achieve excellence in teaching and research.• To generate, disseminate and preserve knowledge.• To meet the challenges of a complex, and modern society through informed social outreach.• To empower through knowledge and information.• To develop a responsible and productive citizenry.• To develop, enhance, and improve the quality of human resources.• To cultivate resolute moral and ethical values.• To meet contemporary regional and national needs and anticipate future social and economic development.• To preserve and promote cultural heritage, humanistic and spiritual values.

Program Outcomes

PO1: Engineering knowledge: Apply the knowledge of mathematics, science, engineering fundamentals, and an engineering specialization to the solution of complex engineering problems.

PO2: Problem analysis: Identify, formulate, research literature, and analyse complex engineering problems reaching substantiated conclusions using first principles of mathematics, natural sciences, and engineering sciences

PO3: Design/development of solutions: Design solutions for complex engineering problems and design system components or processes that meet the specified needs with appropriate consideration for the public health and safety, and the cultural, societal, and environmental considerations.

PO4: Conduct investigations of complex problems: Use research-based knowledge and research methods including design of experiments, analysis and interpretation of data, and synthesis of the information to provide valid conclusions.

PO5: Modern tool usage: Create, select, and apply appropriate techniques, resources, and modern engineering and IT tools including prediction and modelling to complex engineering activities with an understanding of the limitations.

PO6: The engineer and society: Apply reasoning informed by the contextual knowledge to assess societal, health, safety, legal, and cultural issues and the consequent responsibilities relevant to the professional engineering practice.

PO7: Environment and sustainability: Understand the impact of the professional engineering solutions in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.

PO8: Ethics: Apply ethical principles and commit to professional ethics and responsibilities and norms of the engineering practice.

PO9: Individual and team work: Function effectively as an individual, and as a member or leader in diverse teams, and in multidisciplinary settings.

PO10: Communication: Communicate effectively on complex engineering activities with the engineering community and with society at large, such as, being able to comprehend and write effective reports and design documentation, make effective presentations, and give and receive clear instructions.

PO11: Project management and finance: Demonstrate knowledge and understanding of the engineering and management principles and apply these to one's own work, as a member and leader in a team, to manage projects and in multidisciplinary environments.

PO 12: Life-long learning: Recognise the need for, and have the preparation and ability to engage in independent and life-long learning in the broadest context of technological change.

SCHEME OF INSTRUCTION & EXAMINATION
B.E. I- Semester
(ELECTRONICS AND COMMUNICATION ENGINEERING)

S. No.	Course Code	Course Title	Scheme of Instruction				Scheme of Examination			Credits
			L	T	P/ D	Contact Hrs/W	CIE	SEE	Duration in Hrs	
MC: Three Week Induction Programme										
Theory Course										
1	MC 801 PO	Indian Constitution	2	-	-	2	30	70	3	-
2	BS 201 MT	Mathematics-I	3	1	-	4	30	70	3	4
3	BS 202 PH	Engineering Physics	3	1	-	4	30	70	3	4
4	ES 301 EE	Basic Electrical Engineering	3	1	-	4	30	70	3	4
Practical/Laboratory Course										
5	BS 251 PH	Engineering Physics Lab	-	-	3	3	25	50	3	1.5
6	ES 354 EE	Basic Electrical Engineering Lab	-	-	2	2	25	50	3	1
7	ES 353 CE	Engineering Graphics	-	-	3*2	6	50	50	3	3
Total			11	3	11	25	220	430	21	17.5

MC: Mandatory Course**BS:** Basic Science**ES:** Engineering Science**L:** Lecture**T:** Tutorial**P:** Practical**D:** Drawing**CIE:** Continuous Internal Evaluation**SEE:** Semester End Examination (Univ. Exam)**PO:** Political Science**MT:** Mathematics**PH:** Physics**EE:** Electrical Engineering**CE:** Civil Engineering**Note:**

1. Each contact hour is a Clock Hour
2. The duration of the practical class is two clock hours, however it can be extended wherever necessary, to enable the student to complete the experiment

INDIAN CONSTITUTION**MC 801 PO**

Instruction: 2 periods per week

CIE: 30 marks

Credits : -

Duration of SEE: 3 hours

SEE: 70 marks

Objectives:

1. To create awareness among students about the Indian Constitution and To acquaint the working conditions of union, state, local levels, their powers and functions.
2. To create consciousness in the students on democratic values and principles articulated in the constitution and To expose the students on the relations between federal and provincial units.
3. To divulge the students about the statutory institutions.

Outcomes: On successful completion of the course, the students would be able to

1. Know the background of the present constitution of India.
2. Understand the working of the union, state and local levels.
3. Gain consciousness on the fundamental rights and duties.
4. Be able to understand the functioning and distribution of financial resources between the centre and states.
5. Be exposed to the reality of hierarchical Indian social structure and the ways the grievances of the deprived sections can be addressed to raise human dignity in a democratic way.

UNIT – I
Evolution of the Indian Constitution: 1909 Act, 1919 Act and 1935 Act. Constituent Assembly: Composition and Functions; Fundamental features of the Indian Constitution.
UNIT – II
Union Government: Executive-President, Prime Minister, Council of Minister State Government: Executive: Governor, Chief Minister, Council of Minister Local Government: Panchayat Raj Institutions, Urban Government
UNIT – III
Rights and Duties: Fundamental Rights, Directive principles, Fundamental Duties
UNIT – IV
Relation between Federal and Provincial units: Union-State relations, Administrative, legislative and Financial, Inter State council, NITI Ayog, Finance Commission of India
UNIT – V
Statutory Institutions: Elections-Election Commission of India, National Human Rights Commission, National Commission for Women

Suggested Readings:

1.	D.D. Basu, 'Introduction to the constitution of India', Lexis Nexis, New Delhi
2.	Subhash Kashyap, 'Our Parliament', National Book Trust, New Delhi
3.	Peu Ghosh, 'Indian Government & Politics', Prentice Hall of India, New Delhi
4.	B.Z. Fadia & Kuldeep Fadia, 'Indian Government & Politics', Lexis Nexis, New Delhi

MATHEMATICS – I**BS 201 MT***Instruction: 3+1 periods per week**CIE: 30 marks**Credits : 4**Duration of SEE: 3 hours**SEE: 70 marks***Objectives:**

1. To introduce the concepts of sequences, series and their properties
2. To introduce the concepts of functions of several variables and multiple integrals
3. To study vector differential and integral calculus

Outcomes: On successful completion of the course, the students would be able to

1. Find the nature of sequences and series
2. Evaluate multiple integrals
3. Apply this knowledge to solve the curriculum problems
4. Implement the concepts of Multivariable calculus
5. Understand the relevance of vectors to field of engineering.

UNIT – I
Sequences and Series: Sequences, Series, General properties of series, Series of positive terms, Comparison tests, tests of Convergence D'Alembert's ratio test, Cauchy's n^{th} root test, Raabe's test, Logarithmic test, Alternating series, Series of positive and negative terms, Absolute convergence and Conditional convergence.
UNIT – II
Calculus of one Variable: Rolle's theorem, Lagrange's, Cauchy's mean value theorems, Taylor's series, Curvature, Radius of curvature, Circle of curvature, Envelope of a family of curves, Evolutes and Involutives.
UNIT – III
Multivariable Calculus (Differentiation): Functions of two variables, Limits and continuity, Partial derivatives, Total differential and differentiability, Derivatives of composite and implicit functions (Chain rule), Change of variables, Jacobian, Higher order partial derivatives, Taylor's series of functions of two variables, Maximum and minimum values of functions of two variables, Lagrange's method of undetermined multipliers.
UNIT – IV
Multivariable Calculus (Integration): Double integrals, Change of order of integration, Change of Variables from Cartesian to plane polar coordinates, Triple integrals.
UNIT – V
Vector Calculus: Scalar and vector fields, Gradient of a scalar field, Directional derivative, Divergence and Curl of a vector field, Line, Surface and Volume integrals, Green's theorem in a plane, Gauss's divergence theorem, Stoke's theorem (without proofs) and their verification.

Suggested Readings:

1.	R.K. Jain & S.R.K Iyengar, ' <i>Advanced Engineering Mathematics</i> ', Narosa Publications, 2014.
2.	Erwin Kreyszig, ' <i>Advanced Engineering Mathematics</i> ', John Wiley, 9th Edition, 2012.
3.	B.S. Grewal, ' <i>Higher Engineering Mathematics</i> ', Khanna Publications, 43rd Edition, 2014.
4.	G.B. Thomas, Maurice Weir and Joel Hass, Thomas' ' <i>Calculus</i> ', Peterson, 12th Edition, 2010.
5.	B.V. Ramana, ' <i>Higher Engineering Mathematics</i> ', 23rd reprint, 2015.

ENGINEERING PHYSICS**BS 202 PH***Instruction: 3+1 periods per week**CIE: 30 marks**Credits : 4**Duration of SEE: 3 hours**SEE: 70 marks***Objectives:**

1. Aware of limits of classical free electron free theory and to apply band theory of solids
2. Acquire knowledge on various properties of semiconductors.
3. Grasp the intricacies in semiconductor-optical interaction

Outcomes: On successful completion of the course, the students would be able to

1. Distinguish materials based on band theory of solids 3.
2. Classify semiconductors on the basis doping and to estimate conductivity and learn transport phenomenon in semiconductors
3. Appreciate use of optical absorption by semiconductors.
4. Understand the properties of Magnetic materials and superconductivity
5. Understand the operation of LASERS and its applications to optical fibres.

UNIT – I
Crystallography: Introduction, Types of crystal systems, Bravais lattices, Lattice planes and Miller Indices (Cubic system), Inter planar spacing (Cubic system), Bragg's law, Powder diffraction method. Crystal Defects: Classification of point defects, Concentration of Schottky defects in metals and ionic crystals, Concentration of Frankel defects, Line defects, Screw and Edge dislocations, Burger's vector
UNIT – II
Band Theory of Solids & Semiconductors: Classical free electron theory (qualitative), Kronig Penney model (qualitative treatment), Energy band formation in solids, Intrinsic and Extrinsic semiconductors, Concept of a hole, Carrier concentration and conductivity in intrinsic semiconductors, Formation of P-N junction diode and its I – V characteristics, Thermistor and its characteristics, Hall effect and its applications. Dielectric Materials: Dielectrics, Types of polarizations, Electronic, Ionic, Orientational and Space charge polarizations, Expression for Electronic polarizability, Frequency and temperature dependence of dielectric polarizations, Determination of dielectric constant by capacitance Bridge method, Ferro electric, Barium titanate, Applications of Ferroelectrics.
UNIT – III
Wave Mechanics: Matter waves –de-Broglie wavelength, properties of wave function, Physical significance, Schrodinger time dependent and time in-dependent wave equation. Particle in a 1-D box. Electromagnetic theory: Basic laws of electricity and magnetism, Maxwell's equations in integral and differential forms, Conduction and displacement current, Relation between D, E and P – Electromagnetic waves: Equation of plane wave in free space, Poynting theorem.
UNIT – IV
Magnetic Materials: Classification of magnetic materials: dia, para, ferro, antiferro and ferrimagnetic materials, Weiss molecular field theory of ferromagnetism, Magnetic domains, Hysteresis curve, soft and hard magnetic materials, Ferrites: Applications of ferrites. Superconductivity: Introduction, General properties of super conductors, Meissner effect, Type

I and Type II superconductors, BCS theory (qualitative), Introduction to High T _c superconductors, Applications of superconductors.
UNIT – V
<i>Lasers:</i> Characteristics of Lasers, spontaneous and stimulated emission of radiation, Einstein's Coefficients, population inversion, Ruby Laser, Helium Neon Laser, Semi-Conductor Laser and applications of lasers.
<i>Fiber Optics:</i> Introduction, Propagation of light through an optical fiber, Acceptance angle, Numerical aperture (NA), Types of Optical fibers and Refractive index profiles, Fiber drawing process (double Crucible Method), Losses in optical fibers, applications of optical fibers.

Suggested Readings:

1.	B.K. Pandey and S. Chaturvedi, ' <i>Engineering Physics</i> ' Cengage Learning 2012
2.	A.K. Bhandhopadhyaya, ' <i>Nano Materials</i> ', New Age International, 1st Edition, 2007
3.	M.S. Avadhanulu and P.G. Kshirusagar, ' <i>Engg. Physics</i> ', S. Chand & Co. 1st Edition, 1992.
4.	C.M. Srivastava and C. Srinivasan – ' <i>Science of Engg Materials</i> ', New Age International.
5.	R.K Gaur and S.L Gupta- ' <i>Engineering Physics</i> ', Dhanpathrai Publications, New edition.
6.	Sanjay D Jain & Girish G Sahasrabudhe – ' <i>Engineering Physics</i> ', University Press

BASIC ELECTRICAL ENGINEERING**ES 301 EE***Instruction: 3+1 periods per week**Duration of SEE: 3 hours**CIE: 30 marks**SEE: 70 marks**Credits : 4***Objectives:**

1. To provide an understanding of basics in Electrical circuits.
2. To explain the working principles of Electrical Machines and single phase transformers.
3. To apply the principles of Electrical engineering for domestic and industrial safety.

Outcomes: On successful completion of the course, the students would be able to

1. To analyze Electrical circuits to compute and measure the parameters of Electrical Energy.
2. To comprehend the working principles of Electrical DC Machines.
3. To Identify and test various Electrical switchgear, single phase transformers and assess the ratings needed in given application.
4. To comprehend the working principles of electrical AC machines.
5. Appreciate the safety electric installations in domestic and industrial setups

UNIT – I
DC Circuits: Electrical circuit elements (R, L and C), voltage and current sources, Kirchoff current and voltage laws, analysis of simple circuits with dc excitation. Superposition, Thevenin and Norton Theorems.
UNIT – II
AC Circuits: Representation of sinusoidal waveforms, peak and rms values, phasor representation, real power, reactive power, apparent power, power factor. Analysis of single-phase ac circuits consisting of R, L, C, and RL, RC, RLC combinations (series only). Three phase balanced circuits, voltage and current relations in star and delta connections.
UNIT – III
Transformers and 3-ph Induction Motors: Transformers: Electromagnetic induction, Faradays laws, statically induced emf, Lenz law, BH characteristics, ideal and practical transformer, losses and efficiency, Auto-transformer and three-phase transformer connections. Three Phase Induction motor: Generation of rotating magnetic fields, Construction and working of a three-phase induction motor, squirrel cage IM, slip-ring IM, Applications.
UNIT – IV
Single-phase induction motor & DC Machines: Single-phase induction motor: Construction and principle of operation, Capacitor start & capacitor run motor, applications DC Generators: Dynamically induced emf, Flemming's Right hand and Left hand rules, Construction and principle of operation of DC generator, EMF equation, Types of DC Generators, OCC characteristics, applications DC Motors: principle of operation of DC Motor, Types of DC motors, applications.
UNIT – V
Electrical Installations: Components of LT Switchgear: Switch Fuse Unit (SFU), MCB, ELCB, MCCB, Types of Wires and Cables, Earthing. Types of Batteries, Important Characteristics for Batteries. Elementary calculations for energy consumption, power factor improvement and battery backup.

Suggested Readings:

1.	N.K. De, ' <i>Basic Electrical Engineering</i> ', Universities Press, 2015.
2.	J.B. Gupta, ' <i>Fundamentals of Electrical Engineering and Electronics</i> ' S.K. Kataria & Sons Publications, 2002.
3.	J.B. Gupta, ' <i>Utilization of Electric Power and Electric Traction</i> ' S.K. Kataria & Sons Publications, 2010
4.	Abhijit Chakrabarti, Sudipta Nath, Chandan Kumar Chanda, ' <i>Basic Electrical Engineering</i> ' Tata McGraw Hill, Publications, 2009
5.	Hughes, ' <i>Electrical Technology</i> ', VII Edition, International Student -on, Addison Welsey Longman Inc., 1995.

ENGINEERING PHYSICS LAB**BS 251 PH***Instruction: 3 periods per week**CIE: 25 marks**Credits : 1.5**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:**

1. Make precise measurements using basic physical principles and acquire skills to handle the instruments.
2. Relates the theoretical Knowledge to the behavior of Practical Physical world and Analyze errors in the experimental data.
3. Plot graphs between various physical parameters.

Outcomes: On successful completion of the course, the students would be able to

1. Conduct experiments, take measurements independently.
2. Write appropriate laboratory reports.
3. Compute and compare the experimental results and draw relevant conclusions.
4. Use the graphical representation of data and estimate results from graphs
5. Appreciate the application of physical phenomena to the real world.

List of Experiments:

1. To determine the Dielectric constant and Phase transition temperature of Lead Zirconium Titanate (PZT).
2. To draw the I - V Characteristics of P-N Junction diode and to evaluate the resistance.
3. To find the values of Electrical conductivity and energy gap of Ge crystal.
4. Determination of rigidity of modulus of Torsion pendulum.
5. Determination of carrier concentration, Mobility and Hall Coefficient of Ge crystal using Hall Effect Experiment.
6. To determine the constants of A, B and α using Thermistor characteristics.
7. To draw the curve between the magnetizing field and the intensity of magnetization of the specimen (soft iron rod) and to find out i) Coercivity ii) Retentivity and iii) Hysteresis loss.
8. To draw the I - V Characteristics of a solar cell and to calculate the i) Fill factor Efficiency and ii) Series resistance.
9. To Determine the Numerical aperture (NA) of Optical fiber.
10. To determine the wave length of the given Laser source.

Note: Minimum eight experiments should be conducted in the semester**Suggested Readings:**

1. N.K. De, 'Basic Electrical Engineering', Universities Press, 2015.
2. J.B. Gupta, 'Fundamentals of Electrical Engineering and Electronics' S.K. Kataria & Sons Publications, 2002.
3. J.B. Gupta, 'Utilization of Electric Power and Electric Traction' S.K. Kataria & Sons Publications, 2010

BASIC ELECTRICAL ENGINEERING LAB**ES 354 EE***Instruction: 2 periods per week**CIE: 25 marks**Credits : 1**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:**

1. To impart the practical knowledge on testing of DC Machines
2. To impart the practical knowledge on testing of AC
3. To impart the practical knowledge on the usage of common electrical measuring instruments

Outcomes: On successful completion of the course, the students would be able to

1. Get an exposure to common electrical components and their ratings.
2. Analyze the performance of DC and AC Machines.
3. Comprehend the usage of common electrical measuring instruments.
4. Test the basic characteristics of transformers and electrical machines.
5. Appreciate the application of Electrical phenomena to the real world.

Suggested List of Laboratory Experiments/Demonstrations:

- Dem1. Basic safety precautions. Introduction and use of measuring instruments – voltmeter, ammeter, multi-meter, oscilloscope. Real-life resistors, capacitors and inductors.
- Exp 1. Verification of KVL and KCL, superposition theorem (with DC excitation)
- Exp 2 Verification of Thevenins and Nortons theorems (with DC excitation)
- Exp 3. Sinusoidal steady state response of R-L, and R-C circuits – impedance calculation and verification. Observation of phase differences between current and voltage. Power factor calculation.
- Exp 4. Transformers: Observation of the no-load current waveform on an oscilloscope (nonsinusoidal wave-shape due to B-H curve nonlinearity should be shown along with a discussion about harmonics).
- Exp 5. Loading of a transformer: measurement of primary and secondary voltages and currents, and power.
- Exp 6. Three-phase transformers: Star and Delta connections. Voltage and Current relationships (line voltage, phase-to-neutral voltage, line and phase currents).
- Exp 7. Measurement of phase voltage/current, line voltage/current and power in a balanced three-phase circuit connected in star and delta
- Dem2. Demonstration of cut-out sections of machines: dc machine (commutator-brush arrangement), induction machine (squirrel cage rotor), synchronous machine (field winding - slip ring arrangement) and single-phase induction machine.
- Exp 8. OCC characteristics of DC Generator
- Exp 9. Synchronous speed of two and four-pole, three-phase induction motors. Direction reversal by change of phase-sequence of connections.
- Exp 10. Power factor improvement of Induction Motor using static capacitors
- Exp 11. Load Test of DC Motor

Note - 1:

- (i) List of Experiments and Demonstrations suggested above are already available in the Laboratory of the electrical department. No need to purchase any extra equipment except Demonstration 2 equipments
- (ii) Procurement of Demonstration 2 equipments can be done during the course work of that semester. It can be included in the laboratory.

Note - 2:

- (i) Experiments 9, 10 and Demonstration 3 can be incorporated in the Lab syllabus if the topics concerned to the above experiments are considered in new BEE syllabus.

Suggested Readings:

1. N.K. De, '*Basic Electrical Engineering*', Universities Press, 2015.
2. J.B. Gupta, '*Fundamentals of Electrical Engineering and Electronics*' S.K. Kataria & Sons Publications, 2002.
3. J.B. Gupta, '*Utilization of Electric Power and Electric Traction*' S.K. Kataria & Sons Publications, 2010
4. Abhijit Chakrabarti, Sudipta Nath, Chandan Kumar Chanda, '*Basic Electrical Engineering*' Tata McGraw Hill, Publications, 2009
5. Hughes, '*Electrical Technology*', VII Edition, International Student -on, Addison Welsey Longman Inc., 1995.

ENGINEERING GRAPHICS**ES 353 CE***Instruction: 6 periods per week**CIE: 50 marks**Credits: 3**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:**

1. To prepare you to design a system, component, or process to meet desired needs within realistic constraints such as economic, environmental, social, political, ethical, health and safety, manufacturability, and sustainability
2. To prepare you to communicate effectively
3. To prepare you to use the techniques, skills, and modern engineering tools necessary for engineering practice.

Outcomes: On successful completion of the course, the students would be able to

1. Introduction to engineering design and its place in society
2. Exposure to the visual aspects of engineering design
3. Exposure to engineering graphics standards
4. Exposure to solid modeling
5. Exposure to computer-aided geometric design
6. Exposure to creating working drawings
7. Exposure to engineering communication

Sheet No	Description of the Topic	Contact Hours	
		Lecture	Drawing
1.	Principles of Engineering Graphics and their significance, usage of drawing instruments.	1	
2.	Conic Sections – I Construction of ellipse, parabola and hyperbola given focus and eccentricity	1	2
3.	Conic Sections – II Construction of ellipse (given major and minor axis), parabola (given base and height), rectangular hyperbola.		2
4.	Cycloids (cycloid & epicycloid)	1	2
5.	Involutes (involute of triangle, square & circle)		2
6.	Scales (plain & diagonal scales)	1	2+2
7.	Introduction to AutoCAD Basic commands and simple drawings.		2+2
8.	Orthographic Projection Projections of points situated in different quadrants.	1	2
9.	Projections of straight lines – I Line parallel to both the reference planes, line perpendicular or inclined to one reference plane.	1	2
10.	Projections of straight lines – II Line inclined to both the reference planes.	1	2
11.	Projections of planes – I Perpendicular planes	1	2
12.	Projections of planes – II Oblique planes		2
13.	Projections of solids – I Polyhedra and solids of revolution, Projections of solids in simple position.	1	2

14.	Projection of solids – II Projections of solids when the axes inclined to one or both the reference planes.	1	2+2
15.	Section of solids – I When the sectional plane is parallel or perpendicular to one reference plane.	1	2
16.	Section of solids – II When the sectional plane is inclined to one reference plane.		2
17.	Development of surfaces – I Prisms and Cylinders	1	2
18.	Development of surfaces – II Pyramids and Cones		2
19.	Intersection of surfaces – I Intersection of cylinder and cylinder	1	2
20.	Intersection of surfaces – II Intersection of cylinder and cone		2
21.	Isometric projection – I planes and simple solids	1	2
22.	Isometric projection – II combination of two or three solids		2
23.	Conversion of Isometric Views to Orthographic Views	1	2
24.	Floor plans of 2 or 3 rooms including windows, doors, and fixtures such as WC, bath, sink, shower, etc.	1	2

Suggested Text:

1. Bhatt N.D., Panchal V.M. & Ingle P.R., (2014), '*Engineering Drawing*', Charotar Publishing House
2. Shah, M.B. & Rana B.C. (2008), '*Engineering Drawing and Computer Graphics*', Pearson Education
3. S.N Lal, '*Engineering Drawing with Introduction to Auto CAD*', Cengage Learning India Pvt Lid, New Delhi, 2018.
4. Agrawal B. & Agrawal C. M. (2012), '*Engineering Graphics*', TMH Publication
5. Narayana, K.L. & P Kannaiah (2008), '*Text book on Engineering Drawing*', Scitech Publishers
6. (Corresponding set of) CAD Software Theory and User Manuals

NOTE:

1. At least 20 sheets must be covered.
2. Sheet number 1 to 6 (Graph sheets / drawing sheets)
3. Sheet number 7 to 24 (AutoCAD drawings)

SCHEME OF INSTRUCTION & EXAMINATION
B.E. II- Semester
(ELECTRONICS AND COMMUNICATION ENGINEERING)

S. No.	Course Code	Course Title	Scheme of Instruction				Scheme of Examination			Credits
			L	T	P/D	Contact Hrs/W	CIE	SEE	Duration in Hrs	
Theory Course										
1	MC 802 CE	Environmental Science	2	-	-	2	30	70	3	-
2	MC 803 PY	Essence of Indian Traditional Knowledge	2	-	-	2	30	70	3	-
3	HS 101 EG	English	2	-	-	2	30	70	3	2
4	BS 203 MT	Mathematics-II	3	1	-	4	30	70	3	4
5	BS 204 CH	Engineering Chemistry	3	1	-	4	30	70	3	4
6	ES 302 CS	Programming for Problem Solving	3	-	-	3	30	70	3	3
Practical/Laboratory Course										
7	HS 151 EG	English Lab	-	-	2	2	25	50	3	1
8	BS 252 CH	Engineering Chemistry Lab	-	-	3	3	25	50	3	1.5
9	ES 351 CS	Programming for Problem Solving Lab	-	-	2	2	25	50	3	1
10	ES 352 ME	Workshop Practice	-	-	2*3	6	50	50	3	3
Total			15	2	13	30	305	620		19.5

HS: Humanities and Social Sciences**BS:** Basic Science**ES:** Engineering Science**MC:** Mandatory Course**L:** Lectures**T:** Tutorial**P:** Practical**D:** Drawing**CIE:** Continuous Internal Evaluation**SEE:** Semester End Examination (Univ. Exam)**PY:** Philosophy **EG:** English **MT:** Mathematics **CH:** Chemistry**CE:** Civil Engineering, **CS:** Computer Science and Engineering, **ME:** Mechanical Engineering.**Note:**

- Each contact hour is a Clock Hour.
- The students have to undergo a Summer Internship of Rural Agriculture Work Experience (RAW) of one week duration after II-Semester and credits will be awarded in VII semester after evaluation.
- Rural Agriculture Work Experience helps the students primarily to understand the rural situations, status of Agricultural Technologies adopted by farmers and village development plans and to develop skills & attitude of working with farm families for overall development in rural area.
- The main objectives of RAW component are:
 - To make the students familiar with socio-economic conditions of the farmers.
 - To develop communication skills in students using extension teaching methods in transfer of Technology wherever necessary, to enable the student to complete the experiment.

ENVIRONMENTAL SCIENCE**MC 802 CE***Instruction: 2 periods per week**Duration of SEE: 3 hours**CIE: 30 marks**SEE: 70 marks**Credits : -***Objectives:**

1. To create awareness and impart basic knowledge about the environment and its allied problems.
2. To know the functions of ecosystems and To understand importance of biological diversity.
3. To study different pollutions and their impact on environment and To know social and environment related issues and their preventive measures.

Outcomes: On successful completion of the course, the students would be able to

1. Adopt environmental ethics to attain sustainable development.
2. Develop an attitude of concern for the environment.
3. Conservation of natural resources and biological diversity.
4. Creating awareness of Green technologies for nation's security.
5. Imparts awareness for environmental laws and regulations.

UNIT – I
<i>The Multidisciplinary Nature of Environmental Studies:</i> Definition, scope and importance, need for public awareness.
<i>Natural Resources:</i> Water Resources – Use and over utilization of surface and ground water, flood, drought, conflicts over water, Dams: Benefits and Problems. Food Resources –World Food Problems, effects of modern agriculture, fertilizer-pesticides problems, water logging, salinity, Forest Resources – Use and over exploitation, deforestation & its effect on tribal people. Land Resources –Land Degradation, environmental effect of mining, man induced landslides, soil erosion and desertification. Energy Resources –Growing energy needs, Renewable and Non-renewable energy resources
UNIT – II
<i>Ecosystems:</i> Concept of an ecosystem, Structure and function of an ecosystem, Producers, consumers and decomposers, Energy flow in ecosystem, food chains, ecological pyramids, ecological succession, types of ecosystems (marine, pond, river, forest, grassland, desert)
UNIT – III
<i>Biodiversity:</i> Levels of Biodiversity, Bio-geographical classification of India, Value of biodiversity, Threats to biodiversity, endangered and endemic species of India, Conservation of biodiversity, global and national efforts.
UNIT – IV
<i>Environmental Pollution:</i> Definition, Causes, effects and control measures of air pollution, water pollution, soil pollution, noise pollution, thermal pollution, solid waste management.
<i>Environment Protection Act:</i> Air, water, forest and wildlife Acts, issues in the enforcement of environmental legislation.
UNIT – V
<i>Social Issues and the Environment:</i> Watershed management and environmental ethics. Climate change, global warming, acid rain, ozone layer depletion.

Environmental Disaster Management: Types of disasters, impact of disasters on environment, infrastructure, and development. Basic principles of disaster mitigation, disaster management, and methodology. Disaster management cycle and disaster management in India.

Field Work:

- Visit to a local area to document environmental issues- agricultural area/ pond/lake/terrestrial ecosystem
- Visit to a local polluted area- market/slum area/Industrial area/traffic area

Suggested Readings:

1.	A.K. De, ' <i>Environmental Chemistry</i> ', Wiley Eastern Ltd.
2.	E.P. Odum, ' <i>Fundamentals of Ecology</i> ', W.B. Saunders Co., USA.
3.	M.N. Rao and A.K. Datta, ' <i>Waste Water Treatment</i> ', Oxford and IBK Publications.
4.	Benny Joseph, ' <i>Environmental Studies</i> ', Tata McGraw Hill, 2005.
5.	V.K. Sharma, ' <i>Disaster Management</i> ', National Centre for Disaster Management, IIPE, 1999.

ESSENCE OF INDIAN TRADITIONAL KNOWLEDGE**MC 803 PY***Instruction: 2 periods per week**CIE: 30 marks**Credits : -**Duration of SEE: 3 hours**SEE: 70 marks***Objectives:**

1. To get a knowledge in Indian Culture
2. To Know Indian Languages and Literature and the fine arts in India
3. To explore the Science and Scientists of Medieval and Modern India

Outcomes: On successful completion of the course, the students would be able to

1. Understand philosophy of Indian culture.
2. Distinguish the Indian languages and literature.
3. Learn the philosophy of ancient, medieval and modern India.
4. Acquire the information about the fine arts in India.
5. Know the contribution of scientists of different eras.

UNIT – I
Introduction to Culture: Culture, civilization, culture and heritage, general characteristics of culture, importance of culture in human literature, Indian Culture, Ancient India, Medieval India, Modern India
UNIT – II
Indian Languages, Culture and Literature: Indian Languages and Literature-I: the role of Sanskrit, significance of scriptures to current society, Indian philosophies, other Sanskrit literature, literature of south India Indian Languages and Literature-II: Northern Indian languages & literature
UNIT – III
Religion and Philosophy: Religion and Philosophy in ancient India, Religion and Philosophy in Medieval India, Religious Reform Movements in Modern India (selected movements only)
UNIT – IV
Fine Arts in India (Art, Technology & Engineering): Indian Painting, Indian handicrafts, Music, divisions of Indian classic music, modern Indian music, Dance and Drama, Indian Architecture (ancient, medieval and modern), Science and Technology in India, development of science in ancient, medieval and modern India
UNIT – V
Education System in India: Education in ancient, medieval and modern India, aims of education, subjects, languages, Science and Scientists of Ancient India, Science and Scientists of Medieval India, Scientists of Modern India

Suggested Readings:

1.	Kapil Kapoor, 'Text and Interpretation: The India Tradition', ISBN: 81246033375, 2005
2.	'Science in Sanskrit', Samskrita Bharti Publisher, ISBN 13: 978-8187276333, 2007
3.	NCERT, 'Position paper on Arts, Music, Dance and Theatre', ISBN 81-7450 494-X,

	2006
4.	S. Narain, ' <i>Examinations in ancient India</i> ', Arya Book Depot, 1993
5.	Satya Prakash, ' <i>Founders of Sciences in Ancient India</i> ', Vijay Kumar Publisher, 1989
6.	M. Hiriyanna, ' <i>Essentials of Indian Philosophy</i> ', Motilal BanarsidassPublishers, ISBN 13: 978- 8120810990, 2014

ENGLISH**HS 101 EG***Instruction: 2 periods per week**CIE: 30 marks**Credits : 2**Duration of SEE: 3 hours**SEE: 70 marks*

Objectives: To enhance the English language abilities of Engineering students, especially in reading and writing, by

1. Using authentic material for language learning and Exposing them to a variety of content-rich texts
2. Strengthening their grammar and vocabulary and Improving their reading and comprehension skills
3. Honing their writing skills and Encouraging them to think creatively and critically

Outcomes: On successful completion of the course, the students would be able to

1. Read, understand, and interpret a variety of written texts
2. Use appropriate vocabulary and correct grammar
3. Undertake guided and extended writing with confidence.
4. Develop speaking skills and build upon vocabulary
5. Improve overall language skills of SWRL(Speaking, writing, reading and listening)

UNIT – I
Reading: RK Narayan, “A Horse and Two Goats”
Vocabulary: Word formation—Prefixes, Suffixes, Root Words
Grammar: Articles, Prepositions, Determiners
UNIT – II
Reading: Rudyard Kipling, “If”
Vocabulary: Word formation—Compounding and Blending, Contractions
Grammar: Transitions, Connectives
Writing: Paragraph Writing
UNIT – III
Reading: Martin Luther King Jr., “I Have a dream”
Vocabulary: Synonyms, Antonyms, One Word Substitutes
Grammar: Voice
Writing: Letter Writing
UNIT – IV
Reading: Robert Frost, “Road Not Taken”
Vocabulary: Homophones, Homonyms, Homographs
Grammar: Narration (Direct-Indirect Speech)
Writing: Report Writing
UNIT – V
Reading: George Orwell, “The Sporting Spirit” (Excerpt)
Vocabulary: Inclusive Language, Euphemisms
Grammar: Tense
Writing: SOP

Suggested Readings:

1.	Board of Editors. ' <i>Language and Life: A Skills Approach</i> ', Orient Black Swan, 2018.
2.	Sudharshana, NP and C Savitha, ' <i>English for Engineers</i> ', Cambridge University Press, 2018.
3.	Kumar, Sanjay and Pushp Lata, ' <i>English Language and Communication Skills for Engineers</i> ', Oxford University Press, 2018.

MATHEMATICS-II**BS 203 MT***Instruction: 3+1 periods per week**CIE: 30 marks**Credits : 4**Duration of SEE: 3 hours**SEE: 70 marks***Objectives:**

1. To study matrix algebra and its use in solving system of linear equations and in solving eigen value problems
2. To provide an overview of ordinary differential equations and To study special functions like Legendre and Beta Gamma functions
3. To learn Laplace Transforms and its properties

Outcomes: On successful completion of the course, the students would be able to

1. Solve system of linear equations and eigen value problems
2. Solve certain first order and understand the concept of orthogonality
3. Solve higher order differential equations
4. Solve basic problems of Beta Gamma and Legendre's Function.
5. Apply Laplace Transforms; solve ordinary Differential Equations by using it

UNIT – I
Matrices: Rank of a matrix, Echelon form, System of linear equations, Linearly dependence and independence of vectors, Linear transformation, Orthogonal transformation, Eigen values, Eigenvectors, Properties of eigen values, Cayley - Hamilton theorem, Quadratic forms, Reduction of quadratic form to canonical form by orthogonal transformation, Nature of quadratic forms.
UNIT – II
Differential Equations of First Order: Exact differential equations, Integrating factors, Linear differential equations, Bernoulli's, Riccati's and Clairaut's differential equations, Orthogonal trajectories of a given family of curves.
UNIT – III
Differential Equations of Higher Orders: Solutions of second and higher order linear homogeneous equations with constants coefficients, Method of reduction of order for the linear homogeneous second order differential equations with variable coefficients, Solutions of non-homogeneous linear differential equations, Method of variation of parameters, solution of Euler-Cauchy equation.
UNIT – IV
Special Function: Gamma Functions, Beta Functions, Relation Between Beta and Gamma Function, Error Functions. Power Series Method, Legendre's Differential Equations and Legendre's Polynomial $P_n(x)$, Rodrigue's Formula (without proof).
UNIT – V
Laplace Transforms: Laplace Transforms, Inverse Laplace Transforms, Properties of Laplace Transforms and inverse Laplace Transforms, Convolution Theorem (without proof). Solution of ordinary Differential Equations using Laplace Transforms.

Suggested Readings:

1.	R.K. Jain & S.R.K. Iyengar, ' <i>Advanced Engineering Mathematics</i> ', Narosa Publications, 4th Edition, 2014..
2.	Erwin Kreyszig, ' <i>Advanced Engineering Mathematics</i> ', John Wiley, 9th Edition, 2012.
3.	Dr.B.S. Grewal, ' <i>Higher Engineering Mathematics</i> ', Khanna Publications, 43rd Edition, 2014
4.	B.V. Ramana, ' <i>Higher Engineering Mathematics</i> ', 23rd reprint, 2015.
5.	N. Bali, M. Goyal, ' <i>A text book of Engineering Mathematics</i> ', Laxmi publications, 2010
6.	H.K. Dass, Er. Rajnish Varma, ' <i>Higher Engineering Mathematics</i> ', Schand Technical Third Edition.

ENGINEERING CHEMISTRY**BS 204 CH**

Instruction: 3+1 periods per week

Duration of SEE: 3 hours

CIE: 30 marks

SEE: 70 marks

Credits : 4

Objectives:

1. Correlate the properties of materials with their internal structure and use the for Engineering applications and Apply the principals of electrochemistry in storage of electrical energy in batteries and Gain knowledge in causes of corrosion and its prevention.
2. Attains knowledge about the disadvantages of hard water for domestic and industrial purposes. Also learns the techniques of softening of hard water and treatment of water for drinking purpose.
3. Exposed to qualitative and quantitative parameters of chemical fuels and Aware eco-friendly materials and processes.

Outcomes: On successful completion of the course, the students would be able to

1. Apply concept of electrode potential in identifying feasibility of electrochemical reaction; illustrate electro analytical techniques and working of batteries.
2. Identify the mechanism of corrosion of materials on basis of electrochemical approach and devise corrosion control methods.
3. Estimate the physical & chemical parameters of quality of water and explain the process of water treatment.
4. Explain the influence of chemical structure on properties of materials and their choice in engineering applications.
5. Classify chemical fuels and grade them through qualitative analysis.
6. Relate the concept of green chemistry to modify engineering processes and materials.

UNIT – I

Electrochemistry and Battery Chemistry:**Electrochemistry:** Electrochemical cells, Electrolytic and Galvanic cells-notation, cell reaction and cell potentials. Types of electrodes, Calomel Quinhydrone and Glass electrodes. Determination of pH of a solution by using Quinhydrone electrode. Thermodynamics of emf of cells, Nernst equation and its derivation. Applications of Nernst equation to electrode potential and emf of cells. Numerical problems.

Batteries: Primary batteries: Zn - Carbon battery. **Secondary batteries:** Pb-Acid battery and Li-Ion battery, Applications. **Flow batteries (Fuel cells):** Methanol-Oxygen fuel cells, Construction, Applications.

UNIT – II

Water Chemistry and Corrosion: Water Chemistry: Hardness of Water-Types and units of hardness, estimation of temporary and permanent hardness of water by EDTA method. Alkalinity of water and its determination. Water softening by Ion exchange and Reverse Osmosis methods. Numerical problems. Specifications of potable water. Sterilization by Chlorination. Break Point Chlorination.

Corrosion: Causes and its effects. Types of Corrosion-Dry or Chemical corrosion and Wet or Electrochemical corrosion and their mechanism. Electrochemical corrosion –Waterline and Pitting Corrosion. Factors influencing rate of corrosion.

Corrosion control methods: Cathodic protection methods - Sacrificial anodic and impressed current methods. Surface coating methods: Hot Dipping-Galvanizing.

UNIT – III
<p>Engineering Materials: Polymers: Basics of terms polymers: Monomer and its functionality, Polymers and degree of polymerization. Classification of polymers - Thermoplastics & Thermosetting resins. Types of Polymerization (i) Addition (ii) Condensation (iii) Co-Polymerization. Mechanism of free radical polymerization</p> <p>Preparation, Properties & Uses of the following polymers: Plastics - PVC and Bakelite, Fibres - Nylon 6:6, and Kevlar, Elastomers - Buna-S, Butyl and Silicone Rubbers.</p> <p>Conducting polymers : Introduction, Classification and Mechanism of conduction in Poly-acetylene, Applications of conducting polymers.</p> <p>Biodegradable polymers: Introduction preparation, properties and applications of polylactic acid</p>
UNIT – IV
<p>Chemical Fuels: Classification of fuels: Introduction, definition and classification of chemical fuels Primary and secondary fuels. Solid, liquid and gaseous fuels. Requirements of a good fuel. Calorific Value – HCV and LCV. Theoretical calculations of calorific value by Dulong's formula – Numerical problems.</p> <p>Solid Fuels: Coal and its Ranking. Analysis of coal - Proximate and Ultimate analysis.</p> <p>Liquid Fuels: Fractionation of Petroleum. Composition and uses of Gasoline, Diesel and Kerosene. Cracking & its Significance- Catalytic cracking by moving bed method, Knocking. Fuel rating – Octane and Cetane numbers.</p> <p>Gaseous Fuels: LPG, CNG -Composition and Uses.</p> <p>Combustion: Ignition temperature of a fuel, calculation of air quantities by weight and volume required for combustion of a fuel- Numerical problems.</p>
UNIT – V
<p>Green Chemistry and Composites: Green Chemistry: Concept, Principles of green chemistry – Atom Economy, Catalysis. and examples of clean technology.</p> <p>Biodiesel: Sources, Concept of Trans esterification and carbon neutrality. Properties and significance</p> <p>Composites: Introduction to composites, composition and characteristic properties of composites. Classification of composites based on matrix, reinforcement and ply. Applications of composites.</p>

Suggested Readings:

1.	Puri, Sharma and Pathania, ' <i>Principles of Physical Chemistry</i> ', S.N. Chand & Co. New Delhi (Latest edition).
2.	P C Jain and M Jain, ' <i>Engineering Chemistry</i> ', Dhanpat Rai & Sons (15th Edn), New Delhi
3.	J C Kuriacose and J Rajaram, ' <i>Chemistry in Engineering and Technology</i> ', TMH, New Delhi.
4.	O G Palanna, ' <i>Engineering Chemistry</i> ', TMH, and New Delhi.
5.	S S Dara, ' <i>Engineering Chemistry</i> ', S Chand & Sons, New Delhi.
6.	Sashi Chawla, ' <i>Engineering Chemistry</i> ', Dhanpat Rai & Sons, New Delhi.
7.	Shikha Agrawal, ' <i>Engineering Chemistry</i> ', Cambridge, New Delhi.
8.	Prasanta Rath, ' <i>Engineering Chemistry</i> ', Cengage Learning India Pvt. Ltd.

PROGRAMMING FOR PROBLEM SOLVING**ES 302 CS***Instruction: 3 periods per week**Duration of SEE: 3 hours**CIE: 30 marks**SEE: 70 marks**Credits : 3***Objectives:**

1. To introduce the basic concepts of Computing environment, number systems and flowcharts and To familiarize the basic constructs of C language – data types, operators and expressions
2. To understand modular and structured programming constructs in C and To learn the usage of structured data types and memory management using pointers
3. To learn the concepts of data handling using pointers

Outcomes: On successful completion of the course, the students would be able to

1. Formulate simple algorithms for arithmetic and logical problems.
2. Translate the algorithms to programs (in c language).
3. Test and execute the programs and correct syntax and logical errors.
4. Implement conditional branching, iteration and recursion.
5. Decompose a problem into functions and synthesize a complete program using divide and conquer approach.
6. Use arrays, pointers and structures to formulate algorithms and programs.
7. Apply programming to solve matrix addition and multiplication problems and searching and sorting problems.
8. Apply programming to solve simple numerical method problems, namely root finding of function, differentiation of function and simple integration.

UNIT – I
Introduction to Programming: Introduction to components of a computer system (disks, memory, processor, where a program is stored and executed, operating system, compilers etc.). Idea of Algorithm: steps to solve logical and numerical problems. Representation of Algorithm: Flowchart / Pseudocode with examples. From algorithms to programs; source code, variables (with data types) variables and memory locations, Syntax and Logical Errors in compilation, object and executable code.
UNIT – II
Control Structures: Arithmetic expressions and precedence, Conditional Branching and Loops, Writing and evaluation of conditionals and consequent branching. Arrays: Arrays (1-D, 2-D), Character arrays and Strings
UNIT – III
Basic Algorithms: Searching, Basic Sorting Algorithms (Bubble and Selection), Finding roots of Equations. Functions: Functions (including using built in libraries), Parameter passing in functions, call by value. Passing arrays to functions: idea of call by reference
UNIT – IV
Recursion: Recursion, as a different way of solving problems. Example programs, such as Finding Factorial, Fibonacci series. Structure: Structures, Defining structures and Array of Structures

UNIT – V
<i>Pointers</i> - Idea of pointers, Defining pointers, Use of Pointers in self-referential structures, notion of linked list (no implementation), <i>Introduction to File Handling</i> .

Suggested Readings:

1.	Byron Gottfried, ' <i>Schaum's Outline of Programming with C</i> ', McGraw-Hill
2.	A.K. Sharma, ' <i>Computer Fundamentals and Programming in C</i> ', Universities Press, 2nd Edition, 2018.
3.	E. Balaguruswamy, ' <i>Programming in ANSI C</i> ', Tata McGraw-Hill
4.	Brian W. Kernighan and Dennis M. Ritchie, ' <i>The C Programming Language</i> ', Prentice Hall of India.

ENGLISH LAB**HS 151 EG***Instruction: 2 periods per week**CIE: 25 marks**Credits : 1**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:** To enhance the listening and speaking skills of students by

1. Giving them sufficient practice in listening with comprehension and Providing them ample opportunities to improve their public speaking skills
2. Training them in the use of correct pronunciation, stress, and intonation and Sensitizing them to the use of verbal and non-verbal communication appropriate to the context
3. Encouraging them to learn the art of conversation to suit formal and informal situations and Preparing them to make formal presentations and face interviews

Outcomes: On successful completion of the course, the students would be able to

1. Listen, understand, and interpret formal and informal spoken language
2. Speak English with acceptable pronunciation, stress, and intonation
3. Present themselves with confidence in formal situations
4. Participate in individual and group activities with relative ease
5. Improve overall language skills of SWRL(Speaking, writing, reading and listening)

List of Experiments:

1. Listening for Comprehension
2. Pronunciation, Intonation, Stress, and Rhythm
3. Conversation Skills
4. Introducing Oneself and Others
5. Asking for and Giving Information
6. Making Requests and Responding to them Appropriately
7. Giving Instructions and Responding to them Appropriately
8. Making Formal Announcements and Emceeing
9. Group Discussions
10. JAM
11. Role Play
12. Debate
13. Public Speaking Skills and Body Language
14. Interviews
15. Formal Presentations

Suggested Readings:

1. Board of Editors. '*Language and Life: A Skills Approach*', Orient Black Swan, 2018.
2. Balasubramanian, T. '*A Textbook of English Phonetics for Indian Students*'. Macmillan, 1981.
3. '*CIEFL Exercises in Spoken English*', Parts. I-III. Oxford University Press.
4. Pillai, Radhakrishna G. '*Spoken English For You - Level II*', 8th Edition. Emerald Publishers, 2014.
5. Sethi, J and PV Dhamija. '*A Course in Phonetics and Spoken English*', 2nd Edition, Prentice Hall India Learning Private Limited, 1999.

ENGINEERING CHEMISTRY LAB**BS 252 CH***Instruction: 3 periods per week**CIE: 25 marks**Credits : 1.5**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:**

1. Conduct experiments, take measurements and analyze the data through hands-on experience in order to demonstrate understanding of the theoretical concepts of quantitative Analysis while working in small group.
2. Interpret the electro analytical principles with experimental results graphically
3. Demonstrate writing skills through clear laboratory reports

Outcomes: On successful completion of the course, the students would be able to

1. Apply the principles of Colourimetry and Electrochemistry in quantitative estimations.
2. Estimate the rate constants of reactions from concentration of reactants/ products as a function of time.
3. Synthesize small drug molecules.
4. Perform water analysis and comment on the safety of its usage for household applications
5. Appreciate applications of chemical phenomena to real world problems

List of Experiments:

1. Introduction to Chemical Analysis.

2. Techniques of Weighing.

Volumetric Analysis:3. Preparation of Standard Mohr's salt solution, Standardization of KMnO_4 and estimation of ferrous ion.

4. Estimation of Iron(II) by Dichromatometry

Water Analysis:

5. Preparation of Standard Magnesium sulphate solution, standardization of EDTA and Estimation of Total Hardness.

6. Preparation of Standard Sodium Carbonate Solution, Standardization of HCl and Estimation of Carbonate and Bicarbonate Alkalinity.

Conductometry:

7. Estimation of HCl

8. Estimation of CH_3COOH and mixture of acids**Potentiometry**

9. Estimation of HCl

10. Estimation of Iron

pH Metry:

11. Estimation of HCl

Colorimetry:

12. Verification of Beer-Lambert's law and estimation of Manganese.

Chemical Kinetics:

13. Determination of rate constant of acid catalyzed hydrolysis of methyl acetate.

Drug Synthesis

Preparation of Aspirin

Note: Minimum ten experiments should be conducted in the semester

Suggested Readings:

1. B.D. Khosla, A. Gulati and V.Garg, '*Senior Practical Physical Chemistry*', (R. Chand & Co., Delhi)
2. K. K. Sharma and D.S. Sharma, '*An Introduction to Practical Chemistry*', (Vikas publishing, N. Delhi)

PROGRAMMING FOR PROBLEM SOLVING LAB**ES 351 CS***Instruction: 2 periods per week**CIE: 25 marks**Credits : 1**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:**

1. Understand the fundamentals of programming in C Language.
2. Write, compile and debug programs in C and Formulate solution to problems and implement in C.
3. Effectively choose programming components to solve computing problems

Outcomes: On successful completion of the course, the students would be able to

1. Choose appropriate data type for implementing programs in C language.
2. Design and implement modular programs involving input output operations, decision making and looping constructs.
3. Implement search and sort operations on arrays.
4. Apply the concept of pointers for implementing programs on dynamic memory management and string handling.
5. Design and implement programs to store data in structures and files.

Programming Exercise:

1. Finding maximum and minimum of given set of numbers, finding roots of quadratic equation.
2. Sin x and Cos x values using series expansion.
3. Conversion of binary to decimal, octal, hexadecimal and vice versa.
4. Generating Pascal triangle, pyramid of numbers.
5. Recursion: factorial, Fibonacci, GCD.
6. Matrix addition and multiplication using arrays, linear search and binary search using recursive and non-recursive procedures.
7. Bubble sort and selection sort.
8. Programs on pointers: pointer to arrays, pointer to functions.
9. Functions for string manipulations.
10. Programs on structures and unions.
11. Finding the number of characters, words and lines of given text file.
12. File handling programs

Suggested Readings:

1. Byron Gottfried, 'Schaum's Outline of Programming with C', McGraw-Hill
2. A.K. Sharma, 'Computer Fundamentals and Programming in C', Universities Press, 2018.
3. E. Balaguruswamy, 'Programming in ANSI C', Tata McGraw-Hill
4. Brian W. Kernighan and Dennis M. Ritchie, 'The C Programming Language', Prentice Hall of India.

WORKSHOP PRACTICE**ES 352 ME***Instruction: 6 periods per week**CIE: 50 marks**Credits : 3**Duration of SEE: 3 hours**SEE: 50 marks***Objectives:**

1. Identify and use marking out tools, hand tools, measuring equipment and to work to prescribed tolerances and To provide hands on experience about use of different engineering materials, tools, equipments and processes those are common in the engineering field.
2. To gain a good basic working knowledge required for the production of various engineering products.
3. To Study different hand operated power tools, uses and their demonstration and Adopt safety practices while working with various tools

Outcomes: On successful completion of the course, the students would be able to

1. Demonstrate an understanding of and comply with workshop safety regulations.
2. Identify and apply suitable tools for different trades of Engineering processes including drilling, material removing, measuring, chiseling.
3. Study and practice on machine tools and their operations
4. Undertake jobs connected with Engineering Workshop trades including fitting, carpentry, sheet metal, house wiring, welding, smithy and foundry.
5. Apply basic electrical engineering knowledge for house wiring practice

A. TRADE FOR EXERCISES:

1. Carpentry
2. Fitting
3. House wiring
4. Sheet metal working
5. Smithy
6. Welding
7. Plumbing

B. TRADES FOR DEMONSTRATION AND EXPOSURE:

1. Machining (Lathe & Drilling)
2. Injection molding
3. Mould making and casting
4. Basic Electronics lab instruments

C. PRESENTATIONS AND VIDEO LECTURES

1. Manufacturing Methods
2. Rapid Prototyping
3. Glass Cutting
4. 3D printing
5. CNC LATHE

D. IT WORKSHOP: Computer hardware, identification of parts, Disassembly, Assembly of computer to working condition, operating system installation.

Suggested Readings:

1. Venugopal, K, '*Workshop manua*', Anuradha Publications, Kumbakonam, TN, 2012
2. K.C. John, '*Mechanical Workshop*' 2nd Edn., PHI, 2010.
3. Hajra Choudary, '*Elements of Workshop Technology*' Vol. 1, Asian Publishers, Edn., 1993.
4. G.S. Sawhney, '*Mechanical Experiments and Workshop Practice*', I.K. International Publishing House, New Delhi, 2009.

Note: At least two exercises from each trade.